

The spectroscopic characterization of halogenated pollutants through the interplay between theory and experiment: application to R1122

Andrea Pietropolli Charmet ^{1,*} , Giorgia Ceselin ² , Paolo Stoppa ¹  and Nicola Tasinato ^{2,*} 

¹ Dipartimento di Scienze Molecolari e Nanosistemi, Università Ca' Foscari Venezia, Via Torino 155, I-30172 Mestre (Ve), Italy; jacpnike@unive.it

² Scuola Normale Superiore, Piazza dei Cavalieri 7, I-56126 Pisa, Italy; nicola.tasinato@sns.it

* Correspondence: nicola.tasinato@sns.it, N.T.; jacpnike@unive.it, A.P.C.

Abstract: In the last decade halogenated ethenes have seen an increasing interest for different applications, in particular in refrigeration, air-conditioning and heat pumping. At the same time, their adverse effects as atmospheric pollutants require their environmental monitoring, especially by remote sensing spectroscopic techniques. For the purpose, an accurate characterization of the spectroscopic fingerprint, in particular those of relevance for rotational-vibrational spectroscopy, of the target molecules is strongly needed. This work provides an integrated computational-theoretical investigation on R1122 (2-Chloro-1,1-difluoro-ethylene, ClHC=CF₂), a compound widely employed as key intermediate in different chemical processes. State-of-the-art quantum chemical calculations relying on CCSD(T)-based composite schemes and hybrid CCSD(T)/DFT approaches are used to obtain accurate prediction of the structural, rotational and vibrational spectroscopic properties. In addition, the equilibrium geometry is obtained by exploiting the semi-experimental method. The theoretical predictions are used to guide the analysis of the experimentally recorded gas-phase infrared spectrum, which is assigned in the 400 - 6500 cm⁻¹ region. Furthermore, absorption cross sections are accurately determined over the same spectral range. Finally, by using the obtained spectroscopic data, a first estimate of global warming potential of R1122 vibrational spectra is obtained.

Keywords: Quantum chemical calculations; ro-vibrational spectroscopy; environmental chemistry; IR spectroscopy

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1. Introduction

Halogenated ethenes are the shortest members among the family of halogenated olefins, a class of organic compounds which has received growing attention from the scientific community especially in the last decade. Actually, some of them have been proposed as potential and environmentally tolerable alternatives to chlorofluorocarbons (CFCs) in refrigeration, air-conditioning and heat pumping applications. Containing at least one C=C double bond, they possess a reactive site that makes their scavenging from the atmosphere much more efficient. Thus many studies focused on the kinetics and the mechanisms which can be involved in the atmospheric removal processes of these compounds, and so their reactions with OH radicals as well as the ones with O(³P) have been investigated (see, for example [1–7] and references therein). However, it should be noted that some halogenated ethenes are considered potential threats to human health (see, for example [8–10], and references therein), so the assessing of their presence in real-time is clearly desirable, both for experimentally studying their atmospheric chemistry as well as for quantifying their concentrations. The recent advances in high-resolution infrared techniques make them very efficient in monitoring and detecting gaseous pollutants in real-time and with very high sensitivity [11–13], provided that accurate spectroscopic data are available [14]. Generally, these data are obtained by the

37 ro-vibrational and line-shape analysis of high-resolution spectra which are often affected
38 by several resonances [15]; *ab initio* calculations are nowadays able to yield accurate
39 predictions for both anharmonic and Coriolis couplings, thus greatly assisting the whole
40 process.

41 In addition, halogenated ethenes represent well suited probes for characterising
42 the complex balance between van der Waals forces and steric effects that rules inter-
43 molecular interactions. Thus, many works reported on the experimental and theoretical
44 investigations carried out on the heterodimers between halogenated ethenes and noble
45 gases [16–18], acids [19] or other different molecules [20,21].

46 Finally, the size of halogenated ethenes is small enough to make them appropriate
47 test molecules to benchmark the computational predictions obtained at different lev-
48 els of theory, from the state-of-the-art wavefunction-based methods like the coupled
49 cluster with single and double excitations augmented by a perturbational estimate of
50 the effects of connected triple excitations, CCSD(T) [22], to the less computationally de-
51 manding ones rooted in density functional theory (DFT). Within this framework, several
52 investigations (see, for example, [23–29], and references therein) have been carried out
53 to disentangle the anharmonic interactions in their vibrational spectra and to assess the
54 quality of the predicted dipole moment surfaces against the spectroscopic experimental
55 data.

56 Among halogenated ethenes, 2-Chloro-1,1-difluoro-ethylene (R1122, ClHC=CF₂)
57 is currently employed as key intermediate [30] in the production of different relevant
58 compounds like fluorosurfactants, fluorine-containing textile finishing agents, organic
59 silicon fluorine modified resins and other fine chemicals-containing fluorine.

60 Its microwave spectrum has been investigated first by Jenkins *et al.* [31] in the
61 19 – 25 GHz region; almost ten years later, Stone and Flygare [32] extended the analysis
62 in the range 9.9 – 24.1 GHz to determine the off-diagonal terms of ³⁵Cl quadrupole
63 coupling tensor. Subsequently, Leal *et al.* [33] analyzed the range 12 – 225 GHz thus
64 obtaining a set of accurate rotational spectroscopic parameters for both the ³⁵Cl and ³⁷Cl
65 species. Finally, Leung *et al.* [34], by using a combination of broadband chirped pulse
66 and Balle-Flygare cavity Fourier Transform microwave methods, derived the rotational
67 and the quartic centrifugal distortion constants also for the different deuterated and ¹³C
68 isotopologues, as well as the corresponding chlorine nuclear quadrupole coupling tensor.
69 In the same work, the gas-phase structure of the complex between R1122 and Argon has
70 been investigated as well. Very recently, also the structure of the heterodimer involving
71 acetylene has been analyzed [35]. On the other hand, concerning the infrared (IR) spectral
72 region, to the best of our knowledge, the only data available in the literature are the ones
73 reported by Nielsen *et al.* almost seventy years ago [36], where the proposed assignment
74 of its absorption features was carried out mainly by correlating them to those previously
75 assigned for CH₂=CF₂ and CH₂=CCl₂. In addition, there is no information about its
76 absorption cross-section values or its atmospheric lifetime, both of which are mandatory
77 to assess its Global Warming Potential (GWP). Anyway, for the latter Wuebbles *et al.* [37]
78 reported the data of some similar unsaturated halogenated hydrocarbons, and they have
79 maximum lifetime values in the range 10 – 30 days.

80 In the present work we carried out an integrated experimental - computational
81 investigation on the structural and ro-vibrational spectroscopic properties of R1122,
82 whose results can be used to guide further high-resolution analysis as well as to improve
83 the modeling of its atmospheric behaviour. A detailed analysis of the vibrational spectra
84 of this molecule, comprehensive of an accurate determination of the corresponding
85 absorption cross-section data up to 6500 cm⁻¹, was coupled to high-level quantum-
86 chemical calculations performed at different levels of theory. Finally, the accurate values
87 of the integrated band intensities allowed to estimate also the Radiative Forcing (RF)
88 and the corresponding GWP of R1122.

89 2. Materials and Methods

90 2.1. Computational method

91 To accurately characterize the molecular structure and spectroscopic properties of
92 ClHC=CF₂ and to assist the vibrational analysis of experimentally recorded IR spectra,
93 quantum chemical computations were carried out at different levels of theory in order to
94 properly treat both electronic and nuclear problems. The equilibrium structure and the
95 harmonic force field were computed by using the CCSD(T) method in conjunction with
96 medium to large correlation consistent cc-pVnZ basis sets with $n = T, Q$ and 5 [38–40].
97 In order to reduce the one- and N-electron errors, different composite schemes were
98 applied. In particular, the equilibrium structure was computed according to both the
99 geometry gradient scheme [41] and the cheap composite scheme (ChS) [42]. Both of
100 them consider the extrapolation to the complete basis set (CBS) limit and account for
101 core-correlation effects, but the former is formally justified by exploiting the additivity
102 rule to build the gradient used in the geometry optimization, whereas the ChS assumes
103 that the additivity relation can be applied directly on geometrical parameters. On the
104 basis of the available data, both methods are expected to predict bond lengths and
105 valence angles with an accuracy within 2 mÅ and 0.1–0.2°, respectively [43–47]. The
106 ChS was also used for the computation of harmonic frequencies of vibration, as well
107 as for obtaining quartic centrifugal distortion constants and the nuclear quadrupolar
108 coupling constants due to the presence of the Cl nucleus. Within the ChS, the estimate
109 of the target property p^{ChS} (p standing for structural parameters, quartic centrifugal
110 distortion constants, nuclear quadrupolar coupling constants or harmonic vibrational
111 frequencies) is obtained adding on top of the CCSD(T)/cc-pVTZ ansatz corrections that
112 accounts for the CBS extrapolation and core-valence correlation evaluated using the
113 second-order Møller-Plesset (MP2) [48] perturbation theory.

114 Vibrational anharmonic contributions to the computed harmonic properties were
115 evaluated resorting to density functional theory (DFT). According to the recent literature,
116 the hybrid B3LYP [49,50] and PW6B95 [51] functionals in conjunction with a polarized
117 double- ζ basis set, as well as the double-hybrids B2PLYP [52] and rev-DSDPBEP86
118 [53] joined with a triple- ζ basis set can be recommended for the purpose in view of
119 their good performance in the prediction of structural and ro-vibrational spectroscopic
120 properties [54–57]. Specifically, the calendar basis sets [58] jul-cc-pVDZ and jun-cc-pVTZ
121 basis sets were used in conjunction with the PW6B95 and double-hybrid functionals,
122 respectively, while the SNSD [59] basis set was employed in B3LYP calculations. At all
123 the levels of theory considered, geometry optimizations were first carried out, followed
124 by evaluation of analytical Hessians. Cubic and semidiagonal quartic force constants
125 and second- and third-order derivatives of the dipole moment were obtained through
126 numerical differentiation of analytical Hessian matrices, and first-order derivatives of the
127 dipole moment surface, respectively. Double- and triple- ζ basis sets were supplemented
128 by an additional set of d functions on the Cl atom in order to improve the accuracy of the
129 results [55,60]. The relevant spectroscopic parameters were derived in the framework of
130 vibrational perturbation theory to second-order (VPT2) [61–63] by using the computed
131 equilibrium geometries, harmonic properties and anharmonic force constants. Coupled
132 cluster computations were performed by using the CFOUR software [64], while MP2
133 and DFT calculations were carried out employing the Gaussian16 suite of programs
134 [65] which was also adopted for applying VPT2 through its built-in generalized VPT2
135 engine [66,67]. The total energy distribution (TED) analysis of each normal mode was
136 performed by using the INTDER2005 program [68].

137 2.2. Experimental details

138 The gas-phase medium resolution (from 1.0 cm⁻¹ up to 0.2 cm⁻¹) IR spectra of
139 R1122 were recorded in the range of 400 – 6500 cm⁻¹ by employing a Bruker Vertex 70
140 FTIR instrument and using a double walled, stainless steel cell, fitted with KBr windows
141 and with an optical path-length of 134.0 (±0.5) mm. For the vibrational analysis, the

142 spectra were recorded at room-temperature, 128 scans were averaged, and the pressure of
143 the gas was varied in the range of 2.7 – 200 hPa. For the determination of the absorption
144 cross sections, the spectra were obtained at constant temperature (298.0 ± 0.5 K), the
145 pressure of the gas was varied in the range of 2.7 – 112 hPa (N_2 was always added to
146 obtain an overall pressure of 101 kPa, thus minimizing the bias due to finite resolution
147 and instrumental distortion, see [69,70]), and up to 256 interferograms were acquired in
148 order to increase the signal-to-noise ratio. Additional details on the whole experimental
149 procedure can be found in our previous works (see for example [25,47], and references
150 therein).

151 3. Results and discussion

152 R1122 is a planar near-prolate molecule, with asymmetry parameter $\kappa \approx -0.909$,
153 belonging to the C_s symmetry point group. The a and b principal axes of inertia define
154 the molecular plane, while the c axis is perpendicular to it, as illustrated in Figure
155 1. It posses 12 normal modes of vibration that, in terms of symmetry species, can be
156 classified as $9A' \oplus 3A''$; A' vibrations give rise to hybrid a/b bands, while vibrations of
157 A'' symmetry produce c -type absorptions.
158 In the following equilibrium geometries and rotational properties are discussed first,
159 then the attention is moved to the vibrational harmonic force field. Subsequently,
160 anharmonic effects are introduced, and the issuing theoretical predictions used to drive
161 the interpretation of the IR spectra experimentally recorded. This subsection deals also
162 with a description of the main absorption bands and the measurement of integrated
163 absorption cross sections over the $400 - 6500 \text{ cm}^{-1}$ spectral range.

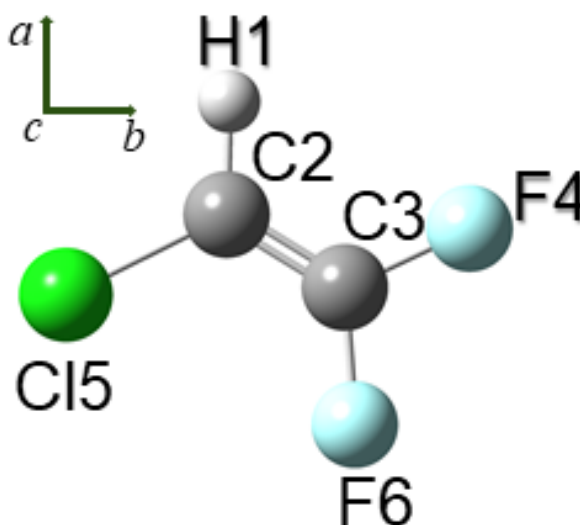


Figure 1. Structure of R1122 with atom labelling. The orientation of the principal axes of inertia is also shown.

164 3.1. Equilibrium geometry and ro-vibrational spectroscopy

165 The fundamental prerequisite for obtaining reliable predictions of spectroscopic
166 parameters, accurate enough to drive the interpretation of experimental spectra, is an
167 accurate molecular geometry, in particular for what concerns rotational spectroscopy.
168 Although the ground-state rotational constants of six isotopologues of ClHC=CF₂ have
169 been experimentally determined [34], to the best of our knowledge, only an effective
170 vibrationally averaged structure (namely the so-called r_0 structure) have been proposed
171 with no attempt to derive the equilibrium geometry. In this respect, it is well known
172 that the semi-experimental (SE) approach is the best method for obtaining accurate
173 equilibrium structures for non-trivial (i.e. two- or three atom-) molecules in the gas-
174 phase [71]. The method makes use of vibrational contributions evaluated theoretically,

175 ΔB^{vib} , to correct the experimentally derived rotational constants, usually those of the
176 ground-vibrational state B_{α}^0 ($\alpha = a, b, c$ being the principal axis of inertia), and obtain the
177 so-called SE equilibrium rotational constants, B_{α}^{SE} , of a set of isotopologues [72]:

$$B_{\alpha}^{\text{SE}} = B_{\alpha}^0 - \Delta B_{\alpha}^{\text{vib}} \quad (1)$$

178 For each isotopologue, vibrational corrections $\Delta B_{\alpha}^{\text{vib}}$ can be computed in the frame-
179 work of VPT2 [61–63]. A non-linear least-squares fitting procedure is then employed to
180 invert the set of SE equilibrium rotational constants to the molecular structural parame-
181 ters, thus returning the SE equilibrium geometry.
182 To obtain the SE equilibrium geometry of R1122, the ground-state rotational constants
183 measured for $^{35}\text{ClHC}=\text{CF}_2$, $^{37}\text{ClHC}=\text{CF}_2$, $\text{ClH}^{13}\text{C}=\text{CF}_2$, $\text{ClHC}=\text{CF}_2$, $^{35}\text{ClDC}=\text{CF}_2$ and
184 $^{37}\text{ClDC}=\text{CF}_2$ [34] and the corresponding vibrational corrections computed at the rev-
185 DSDPBEP86/jun-cc-pV(T+d)Z level of theory have been fed into the MSR software [73]
186 that have been used for the structural refinement. The lack of isotopic substitutions on F
187 atoms, which prevents a reliable determination of the structural parameters involving
188 them, can be overcome by using two strategies: the non-determinable parameters can
189 be constrained to an accurate theoretical value or, alternatively, one can resort to use
190 of the predicate observations method which uses estimates of structural parameters as
191 additional input data [74]. In addition to being more flexible with respect to treating
192 theoretical parameters as rigid constraints, it should lead to more precise results [75,76].
193 In the present work, this second method has been adopted with predicates for the
194 C3–F4, C3–F6 bond lengths and C2C3F4, C2C3F6 angles taken from CCSD(T)/CBS+CV
195 gradient scheme results. The final fit converged to the SE equilibrium geometry reported
196 in Table 1 with a root mean square deviation of $1.7 \times 10^{-3} \text{ u \AA}^2$. All the structural
197 parameters are well determined with maximum standard deviations of 1.1 mÅ for bond
198 lengths and around 0.1° for angles, although for the C3C2H1 angle the 95% confidence
199 interval represents a safer estimate of its accuracy.

200 Indeed, the same table also lists the equilibrium geometry of R1122 obtained from
201 the CCSD(T)/CBS+CV and ChS composite methods and at the CCSD(T)/cc-pV5Z level
202 of theory. The two composite schemes are in perfect agreement with the SE equilibrium
203 geometry, with the largest deviation being 0.9 mÅ and 1.6 mÅ for bond lengths at the
204 CCSD(T)/CBS+CV and ChS level, respectively and within 0.1° for bond angles. The
205 only exception is the C3C2H1 angle which is 0.5° lower accordingly to the theoretical
206 predictions. The equilibrium geometry obtained by using the large cc-pV5Z basis
207 set is also in agreement with the SE structure, even though larger deviations than
208 for composite methods can be noted. A strikingly accurate equilibrium geometry is
209 obtained by augmenting the PW6B95 and rev-DSDPBEP86 results through the nano-
210 LEGO approach [76]. In fact, while for the bare functionals differences as large as 6–7 mÅ
211 are obtained, as it can be seen in Table S1 of the Supplementary Materials (SM), the nano-
212 LEGO procedure lowers the deviations to 1.2 mÅ thus reaching the same accuracy as the
213 CCSD(T)-based composite methods but at a far lower computational cost. Concerning
214 valence angles, the deviations obtained for the C3C2Cl angle at DFT level are somewhat
215 larger than those stemming from CCSD(T)-based approaches, but this is due to the lack
216 of the nano-LEGO parameterization for this angle. By comparing the SE geometrical
217 parameters with the theoretical counterparts obtained by the different methods, it can
218 be speculated that the SE value of the C3C2H1 angle may be overestimated by about
219 0.5° . A possible explanation may be related to imprecision on the rotational constants
220 experimentally determined for the deuterated species, that in turn affects the fitting
221 procedure used for the structural refinement.

222 Moving from the equilibrium geometries, the corresponding equilibrium rotational
223 constants have been derived and then by augmenting them through vibrational correc-
224 tions at DFT level, ground state rotational constants have been obtained. The rotational
225 constants of $^{35}\text{ClHC}=\text{CF}_2$ are compared against the available experimental results in Ta-
226 ble 2 together with quartic centrifugal distortion parameters (data refers to the Watson's

Table 1: Semi-experimental (SE) and theoretical equilibrium geometry of ClHC=CF₂.^a

parameter	SE ^b	CCSD(T)/CBS+CV	ChS	CCSD(T)/V5Z	PW6-nL ^c	rDSD-nL ^d
r(C2–H1)	1.07479(34;79)	1.0753	1.0753	1.0763	1.0756	1.0748
r(C2=C3)	1.3236(11;27)	1.3226	1.3216	1.3259	1.3235	1.3234
r(C3–F4)	1.31421(88;207)	1.3135	1.3157	1.3156	1.3129	1.3140
r(C2–Cl5)	1.71000(71;169)	1.7099	1.7140	1.7152	1.7088	1.7087
r(C3–F6)	1.3073(12;27)	1.3068	1.3084	1.3090	1.3062	1.3071
α(C3C2H1)	120.89(8;20)	120.38	120.64	120.39	120.43	120.29
α(C2C3F4)	123.16(9;22)	123.16	123.29	123.15	123.30	123.19
α(C3C2Cl5)	121.567(41;97)	121.65	121.43	121.65	122.03	121.94
α(C2C3F6)	125.78(6;14)	125.71	125.72	125.72	125.62	125.46

^a Bond lengths in Å, bond angles in deg.

^b Figures in parentheses are standard deviation and 95% confidence intervals in the units of the last significant digits.

^c PW6B95/jul-cc-pV(D+d)Z equilibrium geometry augmented through Nano-LEGO.

^d rev-DSDPBEP86/jun-cc-pV(T+d)Z equilibrium geometry augmented through Nano-LEGO.

227 A-reduction Hamiltonian in the I^r representation) and nuclear quadrupolar coupling
228 constants. The high accuracy obtained in the equilibrium geometry is mirrored in the
229 predicted ground state rotational constants that reproduce the experimental values
230 with a mean absolute percentage error (MAPE) around 0.06% when the composite
231 schemes are considered. The same accuracy is reached by the nano-LEGO corrected
232 rev-DSDPBEP86 functional, which significantly improves the results delivered by the
233 bare functional (MAPE = 0.7%), as it can be appreciated from Table S2 of the SM. A
234 sensible improvement is also noted by comparing PW6B95+nano-LEGO (MAPE = 0.2%)
235 with the corresponding non-augmented counterpart (MAPE = 0.7%), which, in any
236 case, overshoot the accuracy of the rotational constants obtained at the B3LYP/SNSD
237 level of theory (MAPE = 2%, see Table S2 of SM). In passing it is interesting to note
238 that both the CCSD(T)/CBS+CV and ChS composite methods and also the nano-LEGO
239 augmented functionals yield predictions more accurate than the computationally de-
240 manded CCSD(T)/cc-pV5Z level of theory that reproduces the experimental outcomes
241 with a MAPE around 0.4%. Furthermore, by comparing the experimental data [33,34]
242 with the theoretical counterparts, it is noted that the values obtained in ref. [33] appear
243 more accurate than those of ref. [34], which is also coherent with their superior precision,
244 probably because, in the former investigation, the effects of centrifugal distortion have
245 been described up to terms depending on the sixth power of the angular momentum
246 operators, while, in the latter the rotational Hamiltonian has been truncated at the quartic
247 terms. A good agreement can also be reported between experimental and theoretical
248 quartic centrifugal distortion constants obtained from the ChS that, on average achieve
249 the same accuracy (MAPE = 1.3%) as the CCSD(T)/cc-pV5Z computations, but with a
250 significantly lower computational cost. The only notable difference between the two is
251 maximum deviation that amounts to -4.6% and 1.8% for the ChS and CCSD(T)/cc-pV5Z,
252 respectively. The rev-DSDPBEP86 and PW6B95 show slightly larger errors, the MAPE
253 being around 2% and 4% in line with previous benchmark studies [54,55,57].

254 Sextic centrifugal distortion constants have been computed and they are listed in
255 Table 3 where the comparison with the available experimental data [33] is also given
256 again referring to the Watson's A-reduction Hamiltonian in the I^r representation. As it
257 can be seen, the theoretical results obtained at the different levels of theory employed are
258 in overall good agreement, only the PW6B95 seems to yield slightly larger values for the
259 Φ_J and Φ_{JK} centrifugal distortion parameters. Comparing the computed values of Φ_{JK} ,
260 Φ_{KJ} and ϕ_{JK} with the experimental counterparts, a general good agreement can be noted,
261 even though, according to the expected accuracy of the calculations [46,77], the relative
262 deviations between 10 and 15% suggest that the rotational spectra of this molecule may
263 deserve additional investigations, with the aim of extending the analysis toward higher
264 rotational quantum numbers (J in particular). This should lead to the determination of
265 the missing sextic centrifugal distortion parameters which in turn can also affect the
266 values of the remaining ones. In addition, or alternatively, a fit constraining the not yet
267 determined parameters to the theoretical values should be performed. This may avoid
268 the determinable sextic centrifugal distortion constants to be biased in the attempt to
269 account for the centrifugal distortions effects described by the missing parameters. In
270 this regards, the values obtained from the HYB-1 and HYB-2 force fields are suggested
271 for the purpose: the former has been obtained by using the CCSD(T)/CBS+CV geometry,
272 ChS harmonic frequencies (see next subsection) and rev-DSDPBEP86/jun-cc-pV(T+d)Z
273 cubic force constants; the latter has been derived by mixing the CCSD(T)/cc-pV5Z
274 geometry and harmonic force field with CCSD(T)/cc-pVTZ cubic force constants. For
275 the sake of completeness, the rotational spectroscopic parameters of the $^{37}\text{ClHC}=\text{CF}_2$,
276 $^{35}\text{ClHC}=\text{CF}_2$, $^{35}\text{ClH}^{13}\text{C}=\text{CF}_2$, $^{35}\text{ClDC}=\text{CF}_2$ and $^{37}\text{ClDC}=\text{CF}_2$ isotopic species, obtained
277 at different levels of theory, can be found in Tables S.3 - S.7 of the SM.

278 Before concluding this subsection, the α_k^β (k and $\beta = a, b, c$ representing the vibra-
279 tional normal mode and principal axis of inertia, respectively) ro-vibrational interaction
280 constants and the ζ_{kl}^α Coriolis coupling parameters (in absolute value), which are rele-

Table 2: Theoretical rotational-, quartic centrifugal distortion- and nuclear quadrupolar coupling constants of $^{35}\text{ClHC}=\text{CF}_2$ and comparison to experimental values.^a

	CCSD(T)/CBS+CV ^b	ChS ^c	CCSD(T)/V5Z ^d	PW6-nL ^e	rDSD-nL ^f	Exp. ^g	Exp. ^f
A_0	10718.563	10723.797	10680.042	10741.852	10712.614	10710.7335(14)	10710.73661(64)
B_0	2298.164	2298.203	2286.337	2292.253	2295.206	2297.18531(61)	2297.18720(14)
C_0	1891.014	1891.203	1881.791	1887.676	1888.822	1890.14572(36)	1890.14644(15)
MD%	-0.05	-0.07	0.40	0.02	0.05	-	-
MAD%	0.05	0.07	0.40	0.21	0.06	-	-
Δ_J	n.a.	0.350	0.346	0.355	0.339	0.333(12)	0.348727(26)
Δ_{JK}	n.a.	4.26	4.13	4.44	4.077	3.95(11)	4.07532(51)
Δ_K	n.a.	7.75	7.74	7.61	7.69	8.17(11)	7.8803(52)
δ_J	n.a.	0.0596	0.0589	0.06042	0.05790	0.0588(11)	0.059845(8)
δ_K	n.a.	2.60	2.574	2.733	2.544	2.418(97)	2.6008(14)
MD%	n.a.	-0.58	0.76	-2.67	2.12	-	-
MAD%	n.a.	1.39	1.33	4.04	2.14	-	-
χ_{aa}	-54.3	-56.0	-55.3	-51.0	-51.8	-54.8923(48)	-54.81(8)
χ_{bb}	17.7	18.3	18.0	17.7	17.4	18.2356(57)	18.18(4)
χ_{cc}	36.1	37.2	36.7	33.3	34.3	36.6567(56)	36.63(6)
$ \chi_{ab} $	45.7	46.4	46.3	43.0	43.5	47.02(13)	n.a.

^a Rotational parameters within the Watson's A-reduction Hamiltonian in the I' representation. Rotational- and nuclear quadrupolar coupling constants in MHz; quartic centrifugal distortion constants in kHz.

^b Equilibrium rotational constants at CCSD(T)/CBS+CV level corrected through rev-DSDPBEP86/jun-cc-pV(T+d) vibrational contributions. Nuclear quadrupolar coupling constants at CCSD(T)/cc-pVQZ level augmented through rev-DSDPBEP86/jun-cc-pV(T+d) vibrational contributions.

^c Equilibrium rotational constants corresponding to the ChS geometry corrected through rev-DSDPBEP86/jun-cc-pV(T+d) vibrational contributions. Nuclear quadrupolar coupling constants from ChS augmented through rev-DSDPBEP86/jun-cc-pV(T+d) vibrational contributions.

^d Equilibrium rotational constants at CCSD(T)/cc-pV5Z level corrected through rev-DSDPBEP86/jun-cc-pV(T+d) vibrational contributions.

^e Equilibrium rotational constants from Nano-LEGO PW6B95 geometry corrected through PW6B95/jul-cc-pV(D+d)Z vibrational contributions. Centrifugal distortion- and nuclear quadrupolar coupling constants from the bare functional.

^f Equilibrium rotational constants from Nano-LEGO rev-DSDPBEP86 geometry corrected through rev-DSDPBEP86/jun-cc-pV(T+d) vibrational contributions. Centrifugal distortion- and nuclear quadrupolar coupling constants from the bare functional.

^g From Ref. [34].

^f From Ref. [33].

281 vant for further ro-vibrational spectroscopic high-resolution spectroscopic investigations,
 282 are listed in Tables 4 and 5, respectively. Concerning Coriolis resonance, it should be
 283 recalled that, R1122 vibrational levels belonging to the same symmetry species can
 284 interact through *c*-type Coriolis interactions, while levels of different symmetry can be
 285 coupled by both *a*- and *b*-type Coriolis interactions. By using the α_k^β values, vibrational
 286 contributions to rotational constants have been worked out and employed to correct
 287 the equilibrium rotational constants corresponding to the SE structure thus obtaining
 288 the rotational constants of the singly excited fundamental vibrational levels reported in
 289 Table 6.

Table 3: Sextic centrifugal distortion constants (Hz) of $^{35}\text{ClHC}=\text{CF}_2$.^a

	HYB-1 ^b	HYB-2 ^c	CCSD(T)/VTZ	PW6	rDSD	Exp. ^d
$\Phi_J \times 10^5$	6.26	6.65	6.47	7.63	6.03	n.a.
$\Phi_{JK} \times 10^3$	4.20	4.21	4.06	5.62	4.01	n.a.
Φ_{KJ}	-0.0319	-0.0329	-0.0315	-0.0354	-0.0308	-0.0278(24)
Φ_K	0.058	0.058	0.056	0.061	0.056	0.067(15)
$\phi_J \times 10^5$	1.80	1.90	1.85	2.00	1.73	n.a.
$\phi_{JK} \times 10^3$	2.00	2.03	1.96	2.65	1.92	2.25(39)
ϕ_K	0.0909	0.0884	0.0870	0.1025	0.0876	n.a.

^a Watson's A-reduction Hamiltonian in the I' representation.

^b CCSD(T)/CBS+CV geometry, cheap harmonic frequencies and rev-DSDPBEP86/jun-cc-pV(T+d)Z cubic force constants.

^c Geometry and harmonic frequencies at CCSD(T)/cc-pV5Z level and cubic force constants from CCSD(T)/cc-pVTZ computations.

^d From Ref. [33].

Table 4: α_k^β vibrational-rotational interaction constants (MHz) of $^{35}\text{ClHC}=\text{CF}_2$.

Normal mode	<i>a</i>	<i>b</i>	<i>c</i>
1	16.342	2.277	2.014
2	30.201	6.317	4.909
3	31.581	1.337	2.815
4	5.389	1.442	2.336
5	18.068	0.569	0.577
6	-10.680	6.059	5.114
7	0.399	-1.473	-0.048
8	-16.074	1.164	1.833
9	95.383	1.932	1.982
10	6.997	-1.148	-1.999
11	14.983	-0.176	-0.969
12	-76.684	-3.812	-2.423

Table 5: Relevant Coriolis coupling constants of $^{35}\text{ClHC}=\text{CF}_2$.

<i>a</i> -type Coriolis			<i>b</i> -type Coriolis			<i>c</i> -type Coriolis		
Mode <i>k</i>	Mode <i>l</i>	$ \zeta_{kl}^a $	Mode <i>k</i>	Mode <i>l</i>	$ \zeta_{kl}^b $	Mode <i>k</i>	Mode <i>l</i>	$ \zeta_{kl}^c $
1	10	0.973	2	10	0.412	1	2	0.232
1	11	0.173	2	11	0.719	1	3	0.519
2	11	0.478	2	12	0.155	1	4	0.780
1	12	0.119	3	10	0.550	1	5	0.233
3	10	0.110	3	11	0.218	1	6	0.324
3	11	0.750	3	12	0.178	2	3	0.654
4	11	0.232	4	10	0.690	2	4	0.325
5	11	0.219	4	11	0.552	2	5	0.170
6	10	0.171	4	12	0.300	2	6	0.509
6	11	0.211	5	10	0.167	2	7	0.236
6	12	0.471	5	12	0.787	2	8	0.252
7	12	0.543	6	11	0.148	3	4	0.239
8	11	0.171	6	12	0.310	3	5	0.156
8	12	0.371	7	11	0.304	3	7	0.446
9	12	0.551	8	12	0.111	3	8	0.143
			9	11	0.120	4	5	0.128
			9	12	0.355	4	7	0.304
						4	8	0.309
						4	9	0.136
						5	6	0.473
						5	7	0.838
						5	8	0.159
						5	9	0.731
						6	7	0.504
						7	8	0.491
						7	9	0.165
						8	9	0.469

Table 6: Rotational constants (MHz) of the excited fundamental vibrational levels of $^{35}\text{ClHC}=\text{CF}_2$.

Vibrational levels	<i>A</i>	<i>B</i>	<i>C</i>
$v_1 = 1$	10694.453	2294.943	1888.133
$v_2 = 1$	10680.573	2290.896	1885.255
$v_3 = 1$	10679.194	2295.873	1887.323
$v_4 = 1$	10705.395	2295.783	1887.803
$v_5 = 1$	10692.714	2296.652	1889.572
$v_6 = 1$	10721.464	2291.166	1885.045
$v_7 = 1$	10710.372	2298.691	1890.201
$v_8 = 1$	10726.861	2296.053	1888.313
$v_9 = 1$	10615.398	2295.273	1888.163
$v_{10} = 1$	10703.777	2298.361	1892.150
$v_{11} = 1$	10695.802	2297.402	1891.131
$v_{12} = 1$	10787.479	2301.029	1892.570

290 3.2. Harmonic force field

291 Harmonic wavenumbers of $\text{ClHC}=\text{CF}_2$ fundamental vibrations are listed in Table
 292 7 together with the description of the vibrational normal modes based on total energy
 293 distribution (TED) values (%) obtained in terms of the internal coordinates defined in the
 294 lower part of the same Table. For each normal mode the TED analysis has been carried
 295 out in terms of a set of internal coordinates and using the quadratic force constants
 296 obtained at fc-CCSD(T) level of theory. Intensities computed within the double harmonic

297 approximation are also reported, with ChS harmonic intensity of normal mode i , I_i^{ChS} ,
298 computed according to the following expression

$$I_i^{ChS} = I_i^{CCSD(T)/VTZ} + \Delta I_i^{MP2/(T-Q)} + \Delta I_i^{MP2/CV} \quad (2)$$

299 where the first term on the r.h.s. is the harmonic intensity at the CCSD(T)/cc-
300 pV(T+d)Z level, while the second and the third terms account for the enlargement
301 of the basis set and the contribution from the correlation of core electrons, respec-
302 tively. The former is obtained as the difference between MP2 values computed with
303 the cc-pV(Q+d)Z and cc-pV(T+d)Z basis sets, while the latter is the difference between
304 intensities calculated at the MP2/cc-pwCVTZ level by correlating all and only valence
305 electrons. While representing an empirical approximation, the reliability of this ap-
306 proach has been shown to provide reliable predictions [59]. As it can be seen, both
307 frequencies and intensities obtained at the CCSD(T)/cc-pV5Z and ChS levels, are in
308 very good agreement. Indeed, the composite approach, while mostly relying on MP2
309 computations, reproduce the CCSD(T)/cc-pV5Z results with an average accuracy of
310 only 2 cm^{-1} and a maximum differences of 5 cm^{-1} reported for the ω_3 and ω_{10} vibra-
311 tions. Concerning harmonic intensities, the two methods agree on average within 0.7
312 km mol^{-1} , with the largest difference of 3.8 km mol^{-1} observed for the ω_4 vibration,
313 that given the strong intensity of this vibration, in relative terms corresponds to only
314 the 3%. Table 7 also lists the harmonic wavenumbers and intensities of R1122 com-
315 puted at the rev-DSDPBEP86/jun-cc-pV(T+d)Z and PW6B95/jul-cc-pV(D+d)Z levels of
316 theory, while results from B2PLYP/jun-cc-pV(T+d)Z and B3LYP/SNSD computations
317 can be found in Tables S8 and S9 of the SM, respectively. Both rev-DSDPBEP86 and
318 PW6B95 confirms their reliability for computing vibrational properties [55]: in particular,
319 CCSD(T)/cc-pV5Z harmonic frequencies are reproduced with a MAD of only 4.2 cm^{-1}
320 by the rev-DSDPBEP86 double-hybrid functional and 12.8 cm^{-1} by the PW6B95 hybrid
321 functional, while for harmonic intensities the MADs amount to 2.1 and 4.4 km mol^{-1} , re-
322 spectively. As it can be seen by looking at the results reported in Table 7, the TEDs of the
323 ω_5 , ω_6 , ω_7 and ω_8 normal modes show significant contributions from bending and/or
324 stretching involving the chlorine atom; therefore the corresponding $^{35/37}\text{Cl}$ isotopologue
325 splittings should be visible in the experimental spectra.

326 3.3. Vibrational spectroscopy beyond the double harmonic approximation

327 In order to obtain theoretical predictions usable for the quantitative interpretation of
328 experimentally recorded spectra, both mechanical and electrical anharmonicity need to
329 be considered in the calculations. For the purpose, fundamental vibrational frequencies
330 and IR intensities computed beyond the double-harmonic approximation are reported
331 in Table 8, together with the experimentally measured wavenumbers (for graphical
332 comparison between theory and experiment see Figure S1 of the SM), and in Table 9,
333 respectively. While a detailed interpretation of the experimental spectra is deferred to
334 the next subsection, here a comparison among the theoretical outcomes obtained at the
335 different levels of theory is given, pointing out their accuracy with respect to experi-
336 ment and the most important aspects required for the spectral interpretation. Table 8
337 reports the predictions according to four different hybrid force fields: in CC5Z:rDSD and
338 CC5Z:PW6 ones, the harmonic properties from CCSD(T)/cc-pV5Z computations have
339 been mixed with anharmonic contributions evaluated by using the rev-DSDPBEP86 and
340 PW6B95 functionals, respectively; ChS:rDSD and ChS:PW6 are the similar counterparts,
341 but obtained using harmonic frequencies and IR intensities at the ChS level. In addition,
342 the table also collects the predictions from full rev-DSDPBEP86/jun-cc-pV(T+d)Z and
343 PW6B95/jul-cc-pV(D+d)Z computations while B2PLYP and B3LYP results can be found
344 in the supplementary tables S8 and S9, respectively. The different hybrid force fields
345 yield very similar results for the fundamental vibrational frequencies and in general they
346 agree within a few wavenumbers. This is particularly important for the ν_9 , ν_{11} and ν_{12}
347 fundamentals, for which accurate and reliable theoretical predictions become important

Table 7: TED, harmonic wavenumbers (cm^{-1}) and intensities (km mol^{-1}) of $^{35}\text{ClHC}=\text{CF}_2$ normal modes of vibration.

Mode	TED%	Chs		CCSD(T)/V5Z		PW6		rDSD	
		ω	I	ω	I	ω	I	ω	I
ω_1	$R_1(99.5)$	3265	14.34	3261	14.48	3286	17.60	3268	15.64
ω_2	$R_2(72.6) - R_5(9.7) - R_3(7.4)$	1789	172.38	1786	173.84	1821	187.02	1789	185.47
ω_3	$R_5(38.4) - R_6(29.8) - R_3(25.8) + R_9(4.6)$	1365	118.68	1370	120.52	1347	128.80	1362	114.04
ω_4	$R_6(60.2) - R_3(32.3)$	1221	118.61	1224	122.48	1206	128.53	1225	134.28
ω_5	$R_5(31.9) - R_4(23.3) + R_3(19.7) + R_2(13.3)$	991	113.90	991	113.88	991	129.69	993	119.75
ω_6	$R_4(44.1) - R_7(17.0) + R_5(15.2) - R_9(11.3)$	858	8.23	857	8.54	858	10.26	856	9.05
ω_7	$R_8(66.5) - R_3(11.0) - R_7(9.8) + R_9(6.8)$	584	3.04	583	3.20	575	2.78	583	2.96
ω_8	$R_9(64.0) + R_4(25.1) + R_7(7.9) - R_8(6.8)$	437	1.51	436	1.53	434	1.57	437	1.39
ω_9	$R_7(63.3) + R_8(27.4) - R_9(8.2)$	196	1.85	195	1.90	188	1.98	196	1.89
ω_{10}	$R_{12}(82.3) + R_{11}(25.1) + R_{10}(-7.3)$	769	35.88	764	35.30	781	37.68	776	37.21
ω_{11}	$R_{10}(112.0) + R_{12}(-14.5)$	592	0.89	591	0.89	604	0.36	606	0.52
ω_{12}	$R_{11}(72.5) - R_{12}(32.2) - R_{10}(-4.7)$	238	0.59	238	0.62	241	0.70	240	0.54

Internal coordinates definition	
R_1	C–H stretching
R_3	C–F4 stretching
R_5	C–F6 stretching
R_7	C–C–Cl bending
R_9	C–C–F4 bending
R_{11}	H–C–C–Cl out of plane
R_2	C–C stretching
R_4	C–Cl stretching
R_6	C–C–H bending
R_8	C–C–F6 bending
R_{10}	F–C–C–F out of plane
R_{12}	H–C–C–F4 torsion

348 for the assignment of overtones and combination bands, due to the lack of experimental
 349 observations. Indeed, the ν_9 vibration is predicted to occur at quite low wavenumbers,
 350 in a region difficult to access experimentally, whereas the ν_{11} and ν_{12} vibrations, both
 351 having an intensity around 0.5 km mol^{-1} at the anharmonic level, produce absorptions
 352 too weak to be directly detected, even if the ν_{12} frequency has been estimated from the
 353 measurement of difference bands (vide infra). Both the highest levels of theory used
 354 in the present work, i.e. CC5Z:rDSD and ChS:DSD, place ν_9 , ν_{11} and ν_{12} at 195, 580
 355 and 235 cm^{-1} , respectively, and these values have been used for assisting the spectral
 356 interpretation (vide infra). Before moving to the interpretation of the R1122 vibrational
 357 signatures, it is interesting to note that the ν_3 vibration is involved in a Fermi resonance
 358 of type 2 with the $\nu_{10} + \nu_{11}$ combination, which is predicted particularly strong at the
 359 ChS:rDSD level of theory due to the closeness of their deperturbed vibrational energies.
 360 Actually, while this interaction is of weaker magnitude at the other levels of theory
 361 employed, it is described by the following matrix by the ChS:rDSD hybrid force field:

	$\nu_3 = 1$	$\nu_{10} = \nu_{11} = 1$
$\nu_3 = 1$	1333.1	-7.2
$\nu_{10} = \nu_{11} = 1$	-7.2	1334.4

362 whose eigenvalues and eigenvectors are, respectively:

2326	1341
$0.52 \nu_3\rangle$	$0.48 \nu_{10} + \nu_{11}\rangle$
$0.48 \nu_{10} + \nu_{11}\rangle$	$0.52 \nu_3\rangle$

363 As a matter of fact, the $\nu_3 = 1$ and $\nu_{10} = \nu_{11} = 1$ levels are mixed to the same extent
 364 in the perturbed states, thus making an assignment in terms of unperturbed level labels
 365 meaningless. Provided this caveat, the interpretation of the experimental spectrum
 366 remains unaffected and independent of the model hybrid calculation.

367 3.4. Interpreting experiments: vibrational analysis and absorption cross sections

368 The vibrational analysis was performed on the gas-phase spectra measured in the
 369 range $400 - 6500 \text{ cm}^{-1}$. The first step was the assignment of all the strongest fundamen-
 370 tals which was carried out on the spectra obtained at lower pressures; then, the weaker
 371 absorption features (mainly due to overtone and combination bands) were identified in
 372 the spectra obtained at higher pressures. A survey spectrum of the overall region inves-
 373 tigated is reported in Figure 2. Table 8 lists all the assigned fundamentals, together with
 374 the corresponding predicted values obtained at different levels of theory; in the same
 375 way, Table 9 reports the corresponding predicted anharmonic intensities. Finally, Table
 376 10 comprises all the vibrational assignments carried out in the present work together
 377 with the corresponding predicted values, while, the anharmonic constants, x_{ij} derived
 378 from the assigned bands are reported in Table S12 of the SM where, for completeness,
 379 the full list of theoretical values obtained from the CC5Z:rDSD hybrid force field is also
 380 given. For the sake of comparison, the measured fundamental frequencies of R1122 are
 381 juxtaposed with those of similar halogenated ethenes ($\text{H}_2\text{C}=\text{CHCl}$ [78], $\text{H}_2\text{C}=\text{CHF}$ [79],
 382 *cis*- and *trans*- $\text{ClHC}=\text{CHF}$ [80], $\text{ClFC}=\text{CF}_2$ [25]) in Table S13 of the SM.

383 On the basis of quantum chemical calculations, the ν_9 and ν_{12} fundamentals are expected
 384 around 195 and 235 cm^{-1} , respectively. Even though for the latter one the transition
 385 frequency has been experimentally confirmed by the observation of the $\nu_5 - \nu_{12}$ differ-
 386 ence band, the recording of their spectra may require a dedicated investigation through
 387 e.g. the use of synchrotron radiation facilities [81,82]. Furthermore, at room temperature
 388 the two lowest-lying vibrational levels $\nu_9 = 1$ and $\nu_{12} = 1$ present a relative population
 389 of about 39% and 32%, respectively, with respect to the ground vibrational state. For

Table 8: Experimental and theoretical wavenumbers (cm^{-1}) for ClHC=CF₂ fundamental vibrations.

Mode	Exp.	CC5Z:rDSD ^a	CC5Z:PW6 ^b	ChS:rDSD ^c	ChS:PW6 ^d	rDSD	PW6
ν_1	3135.9(3)	3139	3134	3139	3138	3143	3161
ν_2	1747.5(1)	1749	1751	1751	1753	1752	1786
ν_3	1341.7(3)	1342	1348	1341/1326 ^e	1343	1328	1321
ν_4	1200.7(1)	1200	1209	1196	1206	1201	1192
ν_5	971.5(1)/970.2(1) ^f	970	973	972/970 ^f	975/973 ^f	974/972 ^f	975/973 ^f
ν_6	844.9(1)/841.8(5) ^f	843	845	845/842 ^f	847/844 ^f	844/841 ^f	846/844 ^f
ν_7	578.0(1)/577.4 ^f	577	579	578/577 ^f	579/579 ^f	577/577 ^f	570/569 ^f
ν_8	431.8	431	432	432/428 ^f	433/429 ^f	431/427 ^f	430/426 ^f
ν_9	n.a.	195	202	195/193 ^f	202/201 ^f	195/194 ^f	196/194 ^f
ν_{10}	751.1(1)	747	743	753	748	759	762
ν_{11}	n.a.	580	580	581	581	599	594
ν_{12}	n.a.	235	233	236	233	238	237
Max Pos.	-	3.1	8.3	4.0	5.8	8.3	38.5
Max Neg.	-	-3.7	-8.1	-4.4	-2.6	-13.4	-20.9
MD	-	-0.4	1.0	0.4	1.7	0.9	4.1
MAD	-	1.4	3.4	1.5	2.6	4.1	12.0

^a Harmonic frequencies at CCSD(T)/cc-pV5Z level, cubic and quartic force constants at rev-DSDPBEP86/jun-cc-pV(T+d)Z level.

^b Harmonic frequencies at CCSD(T)/cc-pV5Z level, cubic and quartic force constants at PW6B95/jul-cc-pV(D+d)Z level.

^c Harmonic frequencies from cheap composite scheme, cubic and quartic force constants at rev-DSDPBEP86/jun-cc-pV(T+d)Z level.

^d Harmonic frequencies from cheap composite scheme, cubic and quartic force constants at PW6B95/jul-cc-pV(D+d)Z level.

^e $\nu_3/\nu_{10} + \nu_{11}$ frequencies. The two bands are coupled by a Fermi type 2 resonance with equal mixing of the $\nu_3 = 1$ and $\nu_{10} = \nu_{11} = 1$ levels.

^f ³⁵Cl/³⁷Cl.

Table 9: Theoretical anharmonic infrared intensities (km mol^{-1}) for $^{35}\text{ClHC=CF}_2$ fundamental vibrations.

Mode	CC5Z:rDSD ^a	CC5Z:PW6 ^b	ChS:rDSD ^c	ChS:PW6 ^d	rDSD	PW6
ν_1	12.07	11.66	12.00	10.52	13.35	15.06
ν_2	142.05	144.49	132.46	133.64	151.60	161.76
ν_3	84.06	82.18	60.43/45.43 ^e	81.85	93.26	110.61
ν_4	115.04	120.49	111.17	116.09	126.46	122.99
ν_5	84.26	78.91	99.33	97.74	110.50	119.70
ν_6	8.78	8.69	8.52	8.43	8.43	8.61
ν_7	3.13	3.15	2.97	2.99	2.89	2.72
ν_8	1.43	1.41	1.41	1.40	1.30	1.48
ν_9	1.93	1.91	1.88	1.86	1.92	1.99
ν_{10}	34.01	34.39	34.58	34.96	35.93	36.64
ν_{11}	0.86	0.96	0.85	0.95	0.49	0.43
ν_{12}	0.66	0.57	0.63	0.54	0.57	0.65

^a Harmonic intensities at CCSD(T)/cc-pV5Z level augmented by anharmonic contributions at rev-DSDPBEP86/jun-cc-pV(T+d)Z level.

^b Harmonic intensities at CCSD(T)/cc-pV5Z level augmented by anharmonic contributions at PW6B95/jul-cc-pV(D+d)Z level.

^c Harmonic intensities from cheap composite scheme augmented by anharmonic contributions at rev-DSDPBEP86/jun-cc-pV(T+d)Z level.

^d Harmonic intensities from cheap composite scheme augmented by anharmonic contributions at PW6B95/jul-cc-pV(D+d)Z level.

^e $\nu_3/\nu_{10} + \nu_{11}$. The two bands are coupled by a Fermi type 2 resonance with equal mixing of the $\nu_3 = 1$ and $\nu_{10} = \nu_{11} = 1$ levels.

390 this reason, the absorptions due to hot bands contributes significantly to the room tem-
 391 perature IR spectra, even if at the resolutions employed in the present work they are
 392 usually overlapped and covered by the stronger envelopes of the cold bands. If on the
 393 one hand their assignment can be attempted by recording spectra at higher resolutions
 394 (e.g. 0.1 cm^{-1} or better), on the other, the spectral congestion stemming from hot band
 395 absorptions can make high-resolution investigations challenging thus requiring cold
 396 spectra [83].

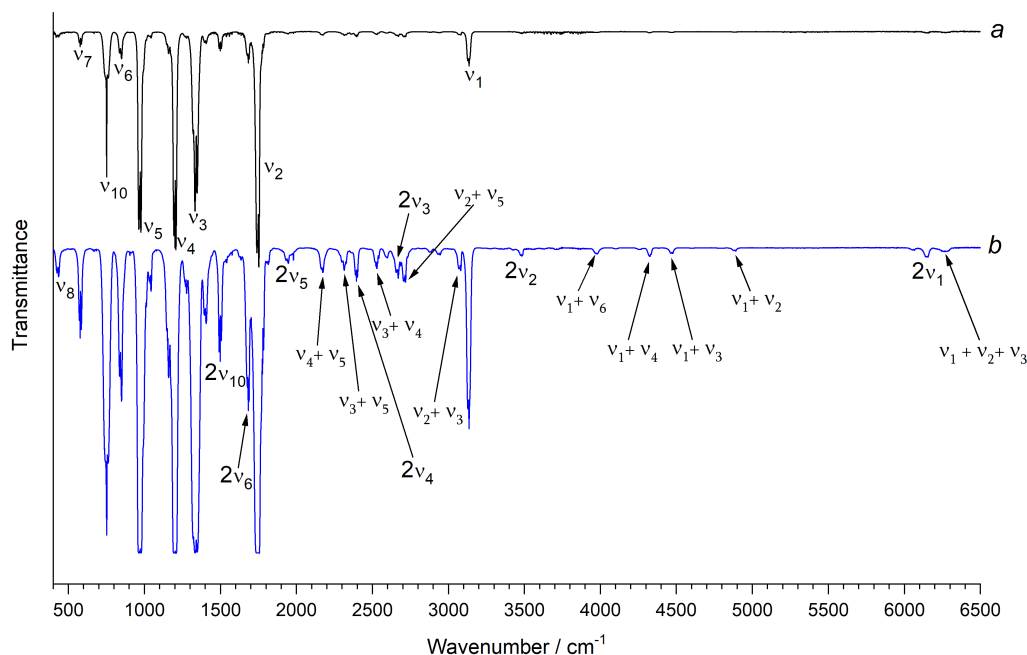


Figure 2. Survey infrared spectra of R1122 in the $400 - 6500\text{ cm}^{-1}$ spectral region. Resolution = 1.0 cm^{-1} , KBr windows, path length = 13.4 cm, room temperature; pressure = 345 Pa (trace a, in black), and 88.77 hPa (trace b, in blue). Only some representative bands are labeled.

397

398 3.4.1. $400 - 800\text{ cm}^{-1}$ spectral region

399 This spectral region is dominated by the strong absorption (with a predicted inten-
 400 sity in the range $34 - 36\text{ km mol}^{-1}$, see Table 9) due to the ν_{10} band (A'' symmetry),
 401 located at 751.1 cm^{-1} , in good agreement with the calculated anharmonic values (ob-
 402 tained at different levels of theory) listed in Table 8. In addition, there are also two weaker
 403 fundamentals of A' symmetry (ν_7 and ν_8) with a computed intensity (at the CC5Z:rDSD
 404 level of theory) of 3.13 and 1.43 km mol^{-1} , respectively; concerning the position of ν_8 ,
 405 there is a very good agreement between its experimental value (at 431.8 cm^{-1}) and the
 406 predicted ones, which are in the range $430 - 432\text{ cm}^{-1}$. Moving to higher wavenumbers,
 407 it is worthwhile to note that the stronger ν_7 , located at 578.0 cm^{-1} , completely obscures
 408 the signals coming from the very weak ν_{11} , predicted in the range $580 - 594\text{ cm}^{-1}$ and
 409 with a computed intensity lower than 1 km mol^{-1} . Focusing on the position of ν_7 all the
 410 theoretical methods listed in Table 8 (with the exception of the PW6B95 hybrid) led to
 411 predicted values having an absolute error not greater than one wavenumber. Besides,
 412 it is worthwhile to note that, even if on the basis of the TED analysis (see Table 7) we
 413 could expect to see the $^{35}/^{37}\text{Cl}$ isotopologue splittings for both these two fundamentals
 414 (i.e., ν_7 and ν_8), only for ν_7 these features are clearly visible in the spectra (signals located
 415 at 578.0 and 577.4 cm^{-1} , respectively), while in the case of ν_8 they were not discernible

Table 10: Vibrational assignment of ClHC=CF₂ and comparison to theoretical wavenumbers (cm⁻¹).

Band	Exp.	CC5Z:rDSD ^a	ChS:rDSD ^b	Band	Exp.	CC5Z:rDSD ^a	ChS:rDSD ^b
ν_8	431.8(3)	431	431	$\nu_2 + \nu_{12}$	1980.6(3)	1983	1986
$2\nu_{12}$	473.5(3)	472	473	$\nu_4 + \nu_5$	2169.3(1)	2167	2165
ν_7	578.0(1)/577.4(1) ^c	577	578/577 ^c	$\nu_3 + \nu_5$	2314.5(5)	2306	2323 ^h
$\nu_5 - \nu_{12}$	736.2	735	736	$\nu_2 + \nu_7$	2323.6(5)	2323	2326
$\nu_9 + \nu_{10} - \nu_9$	747.9(5)	745	753	$2\nu_4$	2394.7(1)	2392	3286
ν_{10}	751.1(1)	747	753	$\nu_5 + 2\nu_9$	2486.9(5)	2486	2479
$\nu_{11} + \nu_{12}$	813.4(5)	815	816	$\nu_3 + \nu_4$	2522.8(1)	2531	2519
ν_6	844.9(1)/842.8(5) ^c	843	845/847 ^c	$\nu_2 + \nu_6$	2599.2(5)	2596	2599
ν_5	971.5(1)/970.2(1) ^c	970	972/970 ^c	$2\nu_5^g$	2663.7(1)	2676	2659
$\nu_{10} + \nu_{12}$	987.8(3)	985	990	$\nu_2 + \nu_5$	2712.8(1)	2713	2716
$\nu_7 + \nu_8$	1007.5(3)	1008	1010	$\nu_2 + \nu_4$	2938.2(3)	2939	2938
$\nu_6 + \nu_9$	1038.4(3)	1037	1038	$\nu_2 + \nu_3$	3074.9(5)	3079	3073
$\nu_6 + \nu_{12}$	1079.1(3)	1078	1080	ν_1	3135.9(3)	3135	3139
$2\nu_7$	1153.1(3)/1152.2(3) ^c	1154	1156/1155 ^c	$\nu_1 + \nu_9$	3325.9(5)	3330	3334
$\nu_5 + \nu_9$	1166.4(3)	1165	1166	$2\nu_2$	3482.7(1)	3484	3489
ν_4	1200.7(1)	1200	1196	$\nu_1 + \nu_7$	3711.9(5)	3712	3717
$\nu_6 + \nu_8$	1273.7(3)	1271	1273	$\nu_1 + \nu_6$	3972.1(5)	3978	3982
$\nu_{10} + \nu_{11}$	1324.9(3)	1325	1326/1341 ^d	$\nu_1 + \nu_5$	4099.8(3)	4104	4109
ν_3	1341.7(3)	1342	1326/1341 ^d	$\nu_2 + \nu_3 + \nu_4$	4261.7(5)	4259	4248
$\nu_5 + \nu_8$	1399.5(3)	1400	1402	$\nu_1 + \nu_4$	4327.8(3)	4328	4328
$2\nu_{10}$	1498.2(1)	1490	1501	$\nu_1 + \nu_6 + \nu_8$	4399.4(5)	4404	4411
$\nu_5 + \nu_7$	1540.1(5)	1545	1547	$\nu_1 + \nu_3$	4471.1(3)	4474	4468
$\nu_3 + \nu_{12}$	1576.4(1)	1577	1568	$\nu_1 + \nu_2$	4884.8(5)	4891	4869
$2\nu_6$	1683.1(1)/1677.5(5) ^c	1680	1682/1676 ^c	$\nu_1 + \nu_3 + \nu_4$	5654.6(5)	5657	5648
ν_2	1747.5(1)	1749	1751	$\nu_1 + \nu_2 + \nu_4$	6058.9(5)	6074	6075
$\nu_4 + \nu_7$	1777.5(1)	1776	1774	$2\nu_1$	6150.1(5)	6157	6166
$\nu_5 + \nu_6$	1813.9(3)	1812	1814	$\nu_1 + \nu_2 + \nu_3$	6218.3(5)	6218	6214
$2\nu_5$	1939.8(3)/1937.3(5) ^c	1938	1940/1938 ^c				

^a Harmonic frequencies at CCSD(T)/cc-pV5Z level, cubic and quartic force constants at rev-DSDPBEP86/jun-cc-pV(T+d)Z level.

^b Harmonic frequencies from cheap composite scheme, cubic and quartic force constants at rev-DSDPBEP86/jun-cc-pV(T+d)Z level.

^c ³⁵Cl/³⁷Cl.

^d $\nu_3/\nu_{10} + \nu_{11}$. The two bands are coupled by a Fermi type 2 resonance with equal mixing of the $\nu_3 = 1$ and $\nu_{10} = \nu_{11} = 1$ levels.

^g Overlapped with $\nu_4 + \nu_{10} + \nu_{11}$ at 2523 cm⁻¹ according to CC5Z:rDSD predictions.

^h Overlapped with $\nu_3 + \nu_{10} + \nu_{11}$ at 2668 cm⁻¹ according to CC5Z:rDSD predictions.

ⁱ $\nu_2 + \nu_{11}$ and $\nu_3 + \nu_{10} + \nu_{12}$ overlapped.

(due to the predominant B-type envelope of this band). The agreement between the experimentally determined and the predicted positions for all the bands assigned in this spectral region can be considered very remarkable (the MAD is only 1.9 and 1.4 cm⁻¹ at the CC5Z:rDSD and ChS:rDSD levels of theory, respectively, see the data listed in Table 10). Finally, it is worthwhile to note that by the $\nu_5 - \nu_{12}$ located at 736.2 cm⁻¹ it is possible to predict the position of ν_{12} , thus obtaining a value of 235.3 cm⁻¹, in a very good agreement with the *ab initio* data of 235 and 236 cm⁻¹ yielded by CC5Z:rDSD and ChS:rDSD levels of theory, respectively (see Table 8).

3.4.2. 800 – 1800 cm⁻¹ spectral region

As expected by the analysis of the data reported in Table 9, in this spectral region the key features (which characterize the spectra measured at lower pressures) are the strong absorptions due to the ν_5 (at 971.5 cm⁻¹), ν_4 (at 1200 cm⁻¹), ν_3 (at 1341.7 cm⁻¹) and ν_2 (at 1747.5 cm⁻¹) fundamentals (all having A' symmetry) which have computed intensities, at the CC5Z:rDSD level of theory, in the range 84 – 142 km mol⁻¹. Located at lower wavenumbers, there is the much weaker ν_6 band (centered at 844.9 cm⁻¹, with a predicted intensity lower than 9 km mol⁻¹). The ν_5 and ν_6 fundamentals show distinct absorption features (located at 971.5/970.2 and 844.9/841.8 cm⁻¹, respectively) due to the presence of both the ³⁵/³⁷Cl isotopologues, thus being in line with the predictions made on the basis of the corresponding TED analysis (as reported in Table 7). Looking at the theoretical data, what is remarkable is the agreement between the CC5Z:rDSD predictions and the corresponding experimental values of these bands (the greatest error is less than 2 wavenumbers). The spectra recorded at increasing pressures allowed the identification of several signals assigned to two-quanta combinations (for example, $\nu_7 + \nu_8$, $\nu_6 + \nu_9$, $\nu_5 + \nu_7$, $\nu_4 + \nu_7$, and so on) as well as to overtone bands ($2\nu_7$, $2\nu_6$ and $2\nu_{10}$). Even in this spectral region, the comparison between the measured positions of the assigned bands (fundamentals and many two-quanta transitions) and the predicted values (listed in Table 10) points out the overall very good performance of the calculations carried out at CC5Z:rDSD level of theory: the corresponding MAD is less than 2 cm⁻¹, while the computed data at ChS:rDSD level of theory has a MAD of 2.5 cm⁻¹.

446
447 3.4.3. 1800 – 3200 cm^{-1} spectral region

448 In the range 1800 – 3200 cm^{-1} only the ν_1 fundamental (A' symmetry, located at
449 3135.9 cm^{-1}) is clearly visible in the spectra recorded at low pressures (as expected by
450 looking at the corresponding calculated intensity, 12.07 km mol^{-1} at the CC5Z:rDSD
451 level of theory, see Table 9). Increasing the sample pressure allowed to identify and
452 assign the signals coming from the $2\nu_5$ (at 1939.8 cm^{-1}), $2\nu_4$ (at 2394.7 cm^{-1}) and $2\nu_3$
453 (at 2663.7 cm^{-1}) overtones as well as several absorptions due to binary combinations
454 mainly involving ν_2 or ν_3 (like $\nu_2+\nu_{12}$, $\nu_3+\nu_5$, $\nu_2+\nu_7$, $\nu_3+\nu_4$, $\nu_2+\nu_5$, $\nu_2+\nu_3$). Looking at the
455 comparison between the experimental and the predicted values (as reported in Table 10)
456 the overall agreement is still very good, the MAD being around 3.5 cm^{-1} for both the
457 CC5Z:rDSD and ChS:rDSD levels of theory.

458
459 3.4.4. 3200 – 6500 cm^{-1} spectral region

460 The analysis of the signals falling in the last spectral region (3200 – 6500 cm^{-1}) was
461 carried out by using the spectra measured at high sample pressure, and several two-
462 and three-quanta combination bands (mainly involving ν_1) were assigned. Besides them,
463 the high-wavenumber side of this region is characterized by the $2\nu_1$ overtone (at 6150.1
464 cm^{-1}) and the nearby $\nu_1+\nu_2+\nu_3$ band. Due to the presence of absorptions involving only
465 two- and (some) three-quanta transitions, this spectral region can be considered more
466 challenging for the theoretical predictions than the former ones; anyway, the overall
467 agreement reached at the CC5Z:rDSD level of theory is very good, being the MAD only
468 4.1 cm^{-1} , whereas in this spectral region the ChS:rDSD predictions present a larger
469 MAD of 9 cm^{-1} . Despite this, the computed anharmonic frequencies and intensities
470 are accurate enough to lead to an unambiguous assignment of the observed spectral
471 features.

472 Using the positions of the fundamentals as benchmark data to assess the perfor-
473 mances of the different anharmonic force fields employed in the present work, the results
474 reported in Table 8 point out the excellent accuracy of CC5Z:rDSD; the MAD is only
475 1.4 cm^{-1} , and all the bands are predicted with errors generally within a few wavenum-
476 bers (the largest absolute deviation is smaller than 4 cm^{-1}). For comparison, all the
477 other composite schemes yielded a bit larger MAD values, even if their predictions can
478 be still considered more than satisfactory; anyway, they all led to bigger deviations,
479 up to around 8 cm^{-1} , for some bands. The very remarkable accuracy offered by the
480 CC5Z:rDSD method is further confirmed by taking into account the whole set of assigned
481 transitions (thus including many overtone and combination bands, in some cases up
482 to three-quanta, see Table 10); the overall MAD is less than 3 cm^{-1} , and most of the
483 deviations are generally lower than 6 cm^{-1} . For comparison, the predictions obtained at
484 ChS:rDSD level of theory have an overall MAD of 4.4 cm^{-1} .

485
486 3.4.5. Absorption cross sections and integrated band intensities

487 In the present work the determination of the absorption cross section spectra of
488 R1122 was carried out by using the medium resolution spectra and following the proce-
489 dure described in details elsewhere (see for example [84] and references therein). Briefly
490 speaking, the method is based on the least-squares fitting the point-by-point absorbance
491 value, $A(\tilde{\nu})$ measured at each wavenumber, $\tilde{\nu}$, versus the corresponding sample con-
492 centration, and always using N_2 as an inert buffer gas. The slope thus obtained at each
493 wavenumber, $\sigma(\tilde{\nu})$, gives the absorbance cross section per molecule ($\text{cm}^2 \text{ molecule}^{-1}$); in
494 the same way also the point-by-point error estimate is obtained (as statistical uncertainty).
495 It is worthwhile to note that this procedure avoids the distortion due to saturation effects
496 and at the same time it leads to better signal-to-noise ratio for the weaker signals; besides,
497 we demonstrated that it produces data in very good agreement with the ones yielded by
498 the line-shape analyses carried out on high-resolution measurements (see, for example,
499 [85–88]).

500 The cross-section spectrum of the overall region investigated is reported in Figure
501 3a, where it is also compared with the theoretical stick spectrum obtained from hybrid
502 CC5Z:rDSD computations (panel b of the same Figure). As it can be seen, the match
503 between the measured and theoretical wavenumbers and relative intensities is very
504 pleasant. In making the comparison, it should be stressed that while the stick spectrum
505 refers to anharmonic IR intensities, the experimental trace represents the cross section
506 spectrum, which yields band intensities upon integrating over a given spectral inter-
507 val as explained above. Indeed, Table 11 lists the experimental integrated absorption
508 cross sections (cm molecule^{-1}) together with the corresponding theoretical predictions
509 obtained at both CC5Z:rDSD and ChS:rDSD levels of theory.

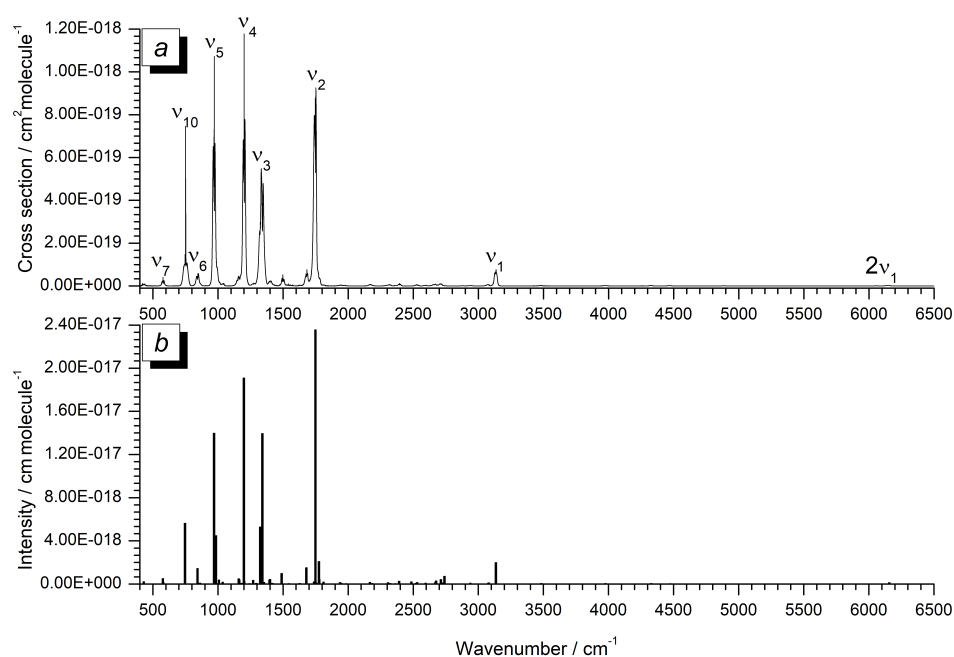


Figure 3. (a) Experimental cross-section spectrum of R1122 in the 400 – 6500 cm^{-1} spectral region. Resolution = 0.2 cm^{-1} , KBr windows, path length = 13.4 cm, 298.0 \pm 0.5 K. Only some representative bands are labeled. (b) Theoretical stick spectrum obtained at the hybrid CC5Z:rDSD level of theory over the same spectral range.

510 In the spectral range 400 – 900 cm^{-1} the ν_{10} band is the most intense absorption;
511 its measured integrated intensity, 5.531(34) $\times 10^{-18}$ cm molecule^{-1} , clearly overcomes
512 the other fundamentals falling in this region (ν_6 has a value of 1.605(10) $\times 10^{-18}$ cm
513 molecule^{-1} , while ν_7 and ν_8 have cross-sections of 6.503(80) and 2.33(27) $\times 10^{-19}$ cm
514 molecule^{-1} , respectively).

515 The strongest absorptions (accounting for more than 68% of the overall integrated
516 integrated band intensities) are localized in the region 900 – 1900 cm^{-1} and they are due
517 to the ν_2 , ν_3 , ν_4 and ν_5 fundamentals, as expected since all their descriptions involve a
518 significant fraction of C-F stretchings (see the corresponding TED% in terms of internal
519 coordinates reported in Table 7). Their individual integrated intensities range from
520 1.8 up to 2.7 $\times 10^{-17}$ cm molecule^{-1} , and their overall sum is equal to 8.50 $\times 10^{-17}$ cm
521 molecule^{-1} . At this point it is worthwhile to note that the overall cross section of R1122
522 in the atmospheric window (900 – 1400 cm^{-1}) can be considered as rather large (around
523 5.9 $\times 10^{-17}$ cm molecule^{-1}), in line to that of other similar halogenated ethenes (for
524 comparison, ClFC=CH₂ has an integrated value of almost 4.5 $\times 10^{-17}$ cm molecule^{-1} ,
525 see [23], while that of ClFC=CF₂ is around 9.3 $\times 10^{-17}$ cm molecule^{-1} , see [25]).

Table 11: Experimental and theoretical integrated absorption cross sections (10^{-19} cm molecule $^{-1}$) of ClHC=CF $_2$ ^a.

Integration limits / cm $^{-1}$	Main absorptions	Exp.	CC5Z:rDSD ^b	ChS:rDSD ^c
400 – 460	ν_8	2.33(27)	2.37	2.46
530 – 620	ν_7	6.503(80)	6.81	6.43
690 – 800	ν_{10}	55.31(34)	56.56	57.52
800 – 890	ν_6	16.05(10)	15.91	15.41
920 – 1080	$\nu_5, \nu_{10} + \nu_{12}$	182.0(16)	190.96	188.41
1090 – 1245	$\nu_4, 2\nu_{11}, \nu_5 + \nu_9, 2\nu_7$	200.4(17)	204.11	209.95
1245 – 1450	$\nu_3, \nu_6 + \nu_8, \nu_{10} + \nu_{11}, \nu_5 + \nu_8$	193.4(13)	209.10	220.31
1450 – 1590	$2\nu_{10}, \nu_5 + \nu_7, \nu_3 + \nu_{12}$	11.24(28)	11.32	11.25
1590 – 1870	$\nu_2, \nu_4 + \nu_7, \nu_5 + \nu_6, 2\nu_6$	263.4(28)	283.61	281.66
1870 – 1970	$2\nu_5$	1.767(34)	2.73	2.34
2090 – 2240	$\nu_4 + \nu_5$	2.546(38)	3.95	3.40
2240 – 2360	$\nu_3 + \nu_5, \nu_2 + \nu_7$	2.69(10)	3.57	2.90
2360 – 2450	$2\nu_4$	2.459(80)	3.58	3.75
2480 – 2560	$\nu_2 + \nu_{10}, \nu_3 + \nu_4$	1.713(37)	4.75	1.92
2560 – 2800	$2\nu_3, \nu_2 + \nu_6, \nu_2 + \nu_5$	7.61(16)	19.87	17.47
2800 – 2990	$\nu_2 + \nu_4$	1.32(15)	1.29	1.39
3020 – 3190	$\nu_1, \nu_2 + \nu_3$	21.41(16)	22.67	22.04
3440 – 3500	$2\nu_2$	0.628(63)	0.84	0.85
4230 – 4350	$\nu_1 + \nu_4$	0.94(20)	1.17	1.16
4430 – 4500	$\nu_1 + \nu_3$	0.61(13)	0.86	0.72
4575 – 4690	$2\nu_2 + \nu_4$	0.045(4)	0.019	0.018
4830 – 4920	$2\nu_1 + \nu_2$	0.303(4)	0.35	0.37
5900 – 6290	$2\nu_1$	1.51(2)	1.71	1.68

^a Values in parentheses are standard errors in the units of the last significant digits. ^b Harmonic frequencies and intensities at CCSD(T)/cc-pV5Z level, anharmonic contributions at rev-DSD-PBEP86-D3/jun-cc-pV(T+d)Z level. ^c Harmonic frequencies and intensities from ChS, anharmonic contributions at rev-DSD-PBEP86-D3/jun-cc-pV(T+d)Z level.

526 The ν_1 fundamental, with an integrated intensity of almost 2.14×10^{-18} cm molecule $^{-1}$,
 527 clearly dominates the region around 3000 cm^{-1} . The other absorptions, falling at higher
 528 wavenumbers, i.e., in the range $3440 - 6500 \text{ cm}^{-1}$, are mainly due to overtone and
 529 combination bands, and their overall integrated cross section is around 4×10^{-19} cm
 530 molecule $^{-1}$.

531 Concerning the comparison between the experimental data and the correspond-
 532 ing computed values, the data reported in Table 11 highlight the overall very good
 533 performance of both CC5Z:rDSD and ChS:rDSD levels of theory in reproducing the
 534 most intense absorptions (i.e., the ones falling in the region $900 - 1900 \text{ cm}^{-1}$), being the
 535 average absolute errors not greater than 6%. The predicted intensities for the weaker
 536 features (like combination and overtone bands) show larger deviations (in line with the
 537 trends seen in previous investigations, see for example [84] and references therein), but
 538 the average absolute error, considering the whole data set of integrated cross-sections
 539 up to 6300 cm^{-1} , is less than 3.5×10^{-19} cm molecule $^{-1}$ (i.e., around 8% of the value of
 540 the overall integrated intensity).

541 By using the obtained cross section spectrum within the narrowband model of Ref.
 542 [89], the radiative forcing (RF) of R1122 has been estimated to be $0.098 \text{ W m}^2 \text{ ppbv}^{-1}$.
 543 Even if we are aware that this approach is not well suited for short-lived molecules and
 544 therefore this determined value should be considered just an estimate of the actual RF,
 545 we note that the data thus computed is in good agreement with the one obtained using
 546 a more sophisticated model on a similar halogenated olefin (e.g. *trans*-1-chloro-3,3,3-
 547 trifluoropropylene, see [37]). As pointed out in the Introduction, for the atmospheric
 548 lifetime of R1122 a reasonable guess of the upper value should be in the range $10 - 30$
 549 days (see also [90]), thus leading to an estimated GWP between 1.5 and 4.5 on a 100-years
 550 time horizon, and between 6 and 18 on a 20-year horizon.

551

552 4. Conclusion

553 Spectroscopic remote sensing techniques are widely used to probe the Earth at-
554 mosphere, to retrieve its composition and to monitor the concentration profiles of a
555 number of species, in particular anthropogenic pollutants which may have hazardous
556 environmental effects or contribute global climate change. In order to exploit the obser-
557 vational data, spectroscopic information need to be accurately determined for the species
558 of potential interest. This represent a huge and time-consuming task that, because of the
559 difficulties in interpreting the experimentally recorded spectra, can be fruitfully achieved
560 by coupling laboratory experiments to state-of-the-art quantum chemical simulations. In
561 the present work, the integrated experimental-theoretical approach to the spectroscopic
562 characterization of atmospheric pollutants has been presented, pointing out the accuracy
563 requirements of quantum chemical calculations for the quantitative interpretation of
564 experimental spectra, using the R1122 molecule as a case study. In particular, a com-
565 prehensive characterization of the structural and rotational/vibrational spectroscopic
566 properties of R1122 has been performed. First, the equilibrium geometry has been de-
567 rived by the semi-experimental approach, in which the ground-state rotational constants
568 of a set of isotopologues have been corrected through vibrational contributions evaluated
569 at the rev-DSDPBEP86/jun-cc-pV(T+d)Z level and used to refine the structural parame-
570 ters in a non-linear least squares procedure. The equilibrium geometry has also been
571 theoretically derived by adopting different methods, in particular CCSD(T)-based com-
572 posite schemes as well as DFT computations relying on the rev-DSDPBEP86 or PW6B95
573 functionals corrected by the recently proposed Nano-LEGO platform. The theoretical
574 geometries have resulted in very good agreement with the SE structure, the deviations
575 being around 1–2 mÅ and 0.2° for bond lengths and angles, respectively. Next, the
576 parameters relevant for rotational spectroscopy, i.e. ground state rotational constants,
577 quartic- and sextic-centrifugal distortion constants and Cl-nuclear quadrupolar coupling
578 constants have then been derived for the different isotopologues of the molecules by
579 means of approaches rooted in CCSD(T) and DFT methods and then compared to the
580 available experimental data. In this respect, the theoretical sextic centrifugal distortion
581 parameters can be used to extend the knowledge of the R1122 rotational fingerprint and
582 to drive the assignment toward high J values; for the isotopologues containing ^{37}Cl , ^{13}C
583 and D atoms sextic distortion parameters are here estimated for the first time. Finally,
584 the vibrational spectroscopic properties have been accurately simulated by accounting
585 for both mechanical and electrical anharmonicity in the framework of VPT2 applied to
586 hybrid force fields in which the harmonic properties derived from the CCSD(T)/cc-pV5Z
587 or ChS computations have been mixed with anharmonic effects evaluated at the rev-
588 DSDPBEP86/jun-cc-pV(T+d)Z level of theory. Moving from the simulated IR spectra, a
589 complete analysis of the experimentally measured gas-phase IR spectra of R1122 in the
590 range 400 – 6500 cm^{-1} has been carried out. The vibrational features have been assigned
591 in terms of fundamentals, overtones, and combination bands up to three quanta, and the
592 corresponding absorption cross-sections have been accurately determined over the same
593 spectral range. Finally, the obtained vibrational spectroscopic data has been employed to
594 obtain the first estimate of the R1122 radiative forcing and, from this, its global warming
595 potential over the 20- and 100-years time horizon.

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597 and N.T.; validation, A.P.C. and N.T.; formal analysis, A.P.C., G.C. and N.T.; investigation, A.P.C.,
598 G.C. and N.T.; resources, A.P.C., N.T. and P.S.; data curation, A.P.C., G.C. and N.T.; writing—
599 original draft preparation, A.P.C. and N.T.; writing—review and editing, A.P.C., P.S., G.C. and
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